

Cycloadditions of Cyclohexynes and Cyclopentyne

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S Supporting Information

ABSTRACT: We report the strategic use of cyclohexyne and the more elusive intermediate, cyclopentyne, as a tool for the synthesis of new heterocyclic compounds. Experimental and computational studies of a 3-substituted cyclohexyne are also described. The observed regioselectivities are explained by the distortion/interaction model.

The study of small rings containing triple bonds has been a topic of vast interest for over 100 years.¹ Following a provocative report in 1902 suggesting the intermediacy of an aryne,² chemists probed the viability of benzyne (**1**, Figure 1)

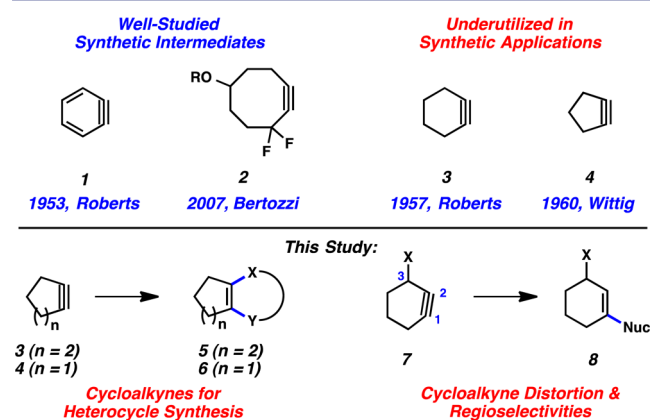


Figure 1. Well-studied cyclic alkynes **1** and **2**, cyclohexyne (**3**), and cyclopentyne (**4**), and objectives of the present study.

and related intermediates. Roberts, in 1953, validated the existence of benzyne (**1**),³ which can be used today in a host of synthetic applications.¹ Perhaps the next most well-studied classes of strained alkynes are cyclooctynes (e.g., **2**) and thiacycloheptynes, which have proven useful in bioorthogonal reactions by Bertozzi, Boons, and others.⁴ In contrast, cyclohexyne (**3**)⁵ and cyclopentyne (**4**)⁶ have seen only sparse use in synthetic applications. Breakthroughs in the manipulation of cyclohexyne include formal C–C bond insertions reported by the laboratories of Stoltz and Carreira,^{7,8} in addition to Diels–Alder reactions as shown by the groups of Guitián and Du Bois.⁹ The use of cyclopentyne has been limited to [2+2] and Diels–Alder cycloadditions.¹⁰

Despite the relatively limited use of **3** and **4** in synthetic applications for the construction of C–C bonds, we envisioned harnessing these strained intermediates to construct new bicyclic heterocyclic scaffolds. Heterocycles are prevalent in

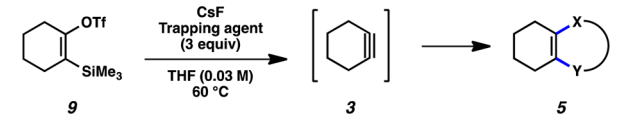
drugs, natural products, and other compounds of tremendous importance.¹¹ Thus, new methods for their synthesis, especially previously inaccessible compounds, remain highly sought after. As suggested in Figure 1, cycloadditions involving **3** or **4** would provide heterocycles **5** or **6**, respectively. Despite the simplicity of this approach, there are no examples of the trapping of cyclohexyne or cyclopentyne to construct heterocycles with one or more newly formed C–X bonds (where X = heteroatom). In addition, we sought to prepare a substituted cyclohexyne **7** and probe regioselectivities in both nucleophilic trapping and cycloaddition reactions. We have previously explained regioselectivities in reactions of substituted benzyne and hetarynes using the distortion/interaction model,¹² but this model has not been tested on the non-aromatic cyclohexyne derivatives. Herein, we demonstrate the synthetic utility of **3** and **4** for the construction of bicyclic heterocycles, and also explain the regioselectivities seen in reactions of the first 3-substituted cyclohexyne using the distortion/interaction model.


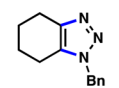



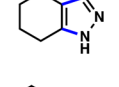

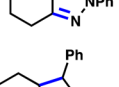

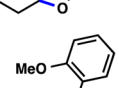

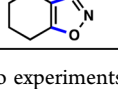
To initiate our study, we opted to generate cyclohexyne in situ from the corresponding silyl triflate, **9** (Table 1). Silyl triflate **9** was first synthesized in 1998,^{9a} but has seen limited use, for example, in Diels–Alder reactions and formal C–C bond insertion reactions.^{7a,9,13,14} We were delighted to find that treatment of silyl triflate **9** with CsF in the presence of a variety of trapping agents delivered heterocyclic products in synthetically useful yields. Specifically, triazoles and pyrazoles were obtained by the trapping of azide and diazo coupling partners, respectively (entries 1–3).¹⁵ An *N*-Ph pyrazole was accessed using a sydnone cycloaddition (entry 4). We also explored nitron and nitrile oxide cycloadditions, which provided isoxazoline- and isoxazole-containing products, respectively (entries 5 and 6). Moreover, additional new trapping experiments to forge 6-membered heterocycles from cyclohexyne are provided in the Supporting Information (SI). It should be emphasized that in contrast to many common methods for heterocycle synthesis, particularly benzyne trapping, the products obtained from cyclohexyne trapping possess more aliphatic character. Being able to access compounds possessing significant sp³ character is an important direction in contemporary drug discovery.¹⁶

Encouraged by our success in building heterocycles from cyclohexyne, we performed trapping experiments of the less well-studied intermediate, cyclopentyne (**4**), using silyl triflate **10** (Table 2).¹⁷ Although silyl triflate **10** has been synthesized previously,¹⁸ no successful trapping experiments involving **10** have been reported. Thus, **10** was treated with CsF in acetonitrile in the presence of various trapping agents. Most

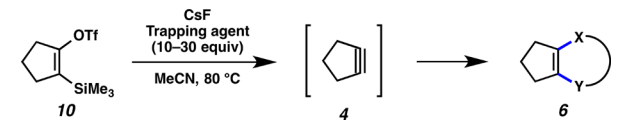
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
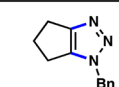

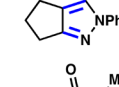

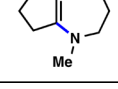
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Table 1. Cycloaddition Reactions of **3** to Construct 5-Membered Heterocycles


Entry	Trapping agent	Product	Yield ^a
1			94% ^b
2			98%
3			82% ^c
4			82%
5			61%
6			90%

^aReported yields are the average of two experiments and are based on the amounts of isolated products. ^bBenzene was used as a cosolvent. ^cEt₂O was used as a cosolvent.

Table 2. Trapping Experiments of Cyclopentyne (**4**)


Entry	Trapping agent	Product	Yield ^a
1			49%
2			59%
3			61%

^aReported yields are the average of two experiments and are based on the amounts of isolated products.

trapping agents gave only low yields or none of the desired products; however, benzyl azide and sydnone partners could be employed to deliver triazole and pyrazole products, respectively (entries 1 and 2). Additionally, we found that trapping of **4** with a cyclic dimethylurea¹⁹ generated a unique product possessing a [5,7]-fused ring system (entry 3).²⁰ Despite the limited scope of trapping agents that can be used to intercept **4**, these studies validate the notion that cyclopentyne can be used in reactions beyond [2+2] and Diels–Alder cycloadditions and may react through non-radical pathways.

Figure 2 shows the optimized structures of cyclohexyne (**3**) and cyclopentyne (**4**) (see the SI for computational details).²¹

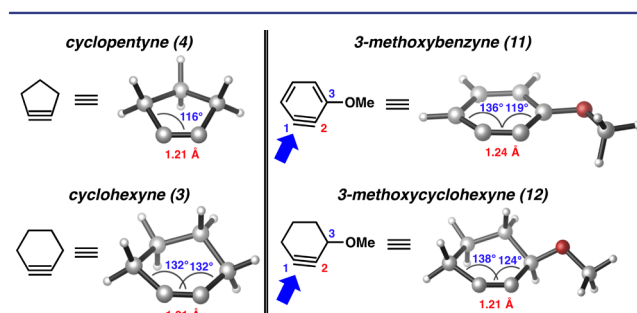


Figure 2. Optimized structures of **3**, **4**, **11**, and **12** obtained using PCM(THF)/M06-2X/6-311+G(2d,p).

The minimum energy conformer of cyclopentyne is slightly puckered and shows C_s symmetry, in agreement with previous studies.²² The significant angle-strain of this structure is revealed by the large deviation of the internal ring angles (116°) from the ideal linear disposition of alkynes. The strain has been calculated to be ca. 74 kcal mol⁻¹.²² Such a large strain distorts the in-plane π bond in a way that cyclopentyne has ~10% calculated diradical character.²³ The more relaxed internal angle in cyclohexyne (132°) causes a smaller, but still significant strain, estimated as ca. 44 kcal mol⁻¹.²² Cyclohexyne (**3**) possesses a C₂-symmetric structure that resembles the well-known half-chair structure of cyclohexene.^{22a,24}

We also compared 3-methoxybenzene (**11**) to its non-aromatic counterpart, 3-methoxycyclohexyne (**12**) (Figure 2). In the case of **11**, as we have previously described,^{12a,b} the inductively withdrawing methoxy group at C3 distorts the aryne significantly. Nucleophilic trapping occurs at C1, the more linear aryne terminus whose reactive orbital possesses more *p* character, uniformly with high degrees of regioselectivity. Interestingly, 3-methoxycyclohexyne (**12**) bears similar distortion, with internal angles at C1 and C2 being calculated as 138° and 124°, respectively.²⁵ Much like the distortion seen in **11**, the distortion in **12** is attributed to the inductively withdrawing nature of the C3 methoxy group that causes rehybridization of C2 (Bent's rule, see SI for further discussion).²⁶ Consequently, we predict that nucleophilic addition to 3-alkoxycycloalkynes should occur with a significant preference for attack at C1, the more linear alkyne terminus.

To test our prediction, we prepared benzyloxysilyl triflate **13**, the first C3-substituted cyclohexyne precursor, and performed trapping experiments (Figure 3).²⁷ When silyl triflate **13** was treated with CsF in the presence of imidazole, adduct **15** was obtained exclusively, which arises via attack at C1 of cyclohexyne **14**. Similarly, trapping with benzyl azide gave a 5.1 to 1 ratio of cycloadducts **16** and **17**, which is consistent with a preference for initial bond formation occurring between the more nucleophilic terminus of the azide²⁸ and C1 of **14**.

Figure 4 shows the calculated competing transition states, TS1–TS4, for the reactions shown in Figure 3.²⁹ In agreement with the observed selectivity, attack by imidazole at C1 is highly preferred (by ca. 3 kcal mol⁻¹) because 3-methoxycyclohexyne (**12**) is pre-distorted toward the preferred transition state, TS1. Similarly, in the azide cycloaddition, TS3 is favored over TS4, although the calculated regioselectivity is overestimated. It is important to note the systematic increase in distortion energy (Δ*E*^{dist}), the cost of altering the substrate geometry toward the

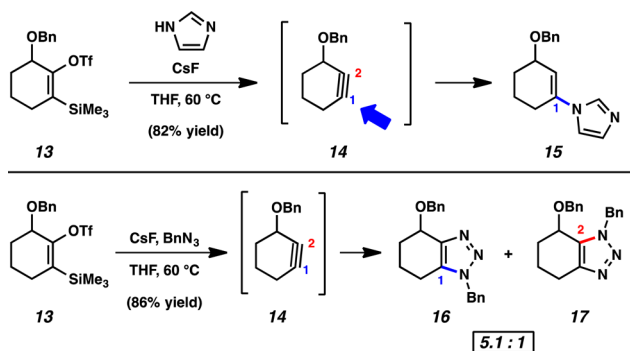


Figure 3. Experimental results validate regioselectivity predictions in reactions of 3-substituted cyclohexyne **14**.

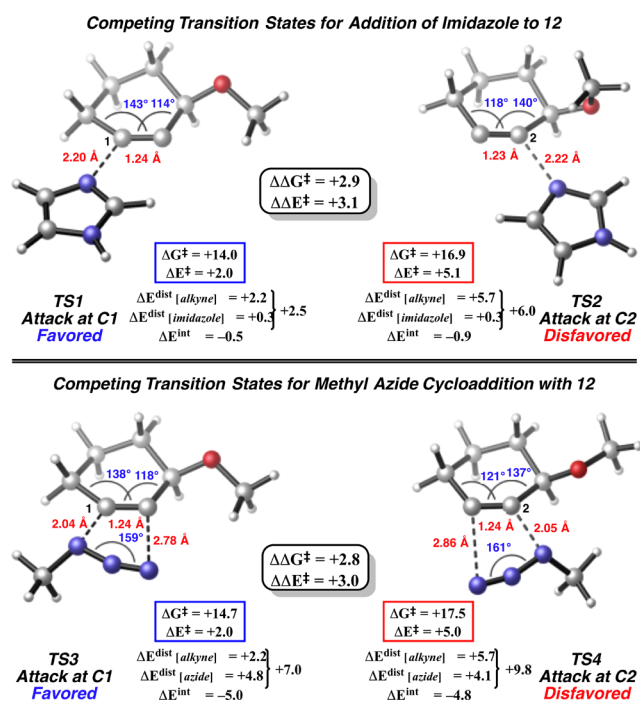
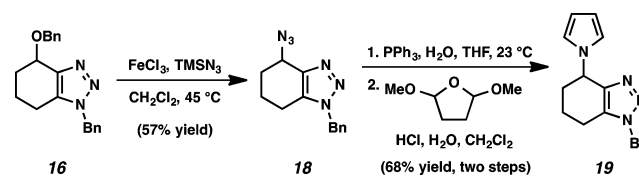


Figure 4. Optimized transition states for nucleophilic addition by imidazole and cycloaddition with methyl azide to **12** using PCM(THF)/M06-2X/6-311+G(2d,p). Energies are provided in kcal mol⁻¹.

transition state, of the 3-methoxycyclohexyne moiety in **TS2** and **TS4**, which accounts for most of the calculated energy differences between competing transition states. As a common feature of distortion-controlled reactions, the interaction energies (ΔE^{int}), or in other words, the stabilization due to orbital overlap between the reacting fragments in the transition state, is nearly identical when comparing competing transition states. It is notable that this trend is observed for both the imidazole and azide trapping agents, despite their different electronic properties.

In addition to serving as a probe to assess the distortion/interaction model, benzyloxycyclohexyne **14** can also be used to access highly functionalized heterocycles. As shown in Scheme 1, triazole **16**, prepared from the benzylazide cycloaddition of **14** (Figure 3), was converted to azide **18** through an uncommon functionalization of a pseudobenzyl benzyloxy group.³⁰ Subsequent reduction and pyrrole formation provided triazolopyrrole **19**.

Scheme 1. Elaboration of Benzyloxycyclohexyne **16** to Triazolopyrrole **19**



In summary, we have demonstrated that cyclohexyne and the more elusive intermediate, cyclopentyne, serve as effective tools for the synthesis of new heterocyclic compounds. We have also shown that the distortion/interaction model correctly predicts regioselectivities in reactions of the first 3-substituted cyclohexyne. This validates the distortion/interaction model as a powerful predictive tool for gauging cycloalkyne regioselectivities, just from the reactant's structure, while also providing the impetus for the further exploration of highly strained cycloalkynes as valuable synthetic building blocks.

■ ASSOCIATED CONTENT

Supporting Information

Detailed experimental and computational procedures, compound characterization, Cartesian coordinates, electronic energies, entropies, enthalpies, Gibbs free energies, and lowest frequencies of the calculated structures. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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Notes

The authors declare no competing financial interest.

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